

# A new rearrangement of spiro[indane-1,3'-thiophene] and spiro[naphthalene-1,3'-thiophene] derivatives accompanied by opening of the cycloalkane ring

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**Abstract** The new thioamides 2-oxoindane-1-carbothioamides and 2-oxo-1,2,3,4-tetrahydronaphthalene-1-carbothioamides underwent conjugated addition to (*E*)- $\beta$ -nitrostyrenes followed by cyclization to give products containing 5-hydroxyimino-2-aryliminothiophene rings spiro-annulated to the 2-indanone or 2-tetralone system. On treatment with hydrochloric acid in boiling methanol the compounds underwent a new rearrangement, involving transformation of the 2-aryliminothiophene ring to a pyrrole and an opening of the alicyclic ring in the proximity of the carbonyl group, affording 2-(1,3-diaryl-2-oxo-5-thioxopyrrol-4-yl)benzeneacetic or 2-(1,3-diaryl-2-oxo-5-thioxopyrrol-4-yl)benzenepropanoic acid methyl esters. The structures of two selected pyrrole derivatives were determined by X-ray crystal analyses. The thioamides derived from 2-indanone showed unexpected reactivity towards primary alcohols and underwent conversion to 2-hydroxy-1*H*-indene-3-thiocarboxylic acid *O*-esters.

**Keywords** Heterocycles · Michael addition · Thioamides · Nitrostyrenes · Spiro compounds · X-ray structure determination

## Introduction

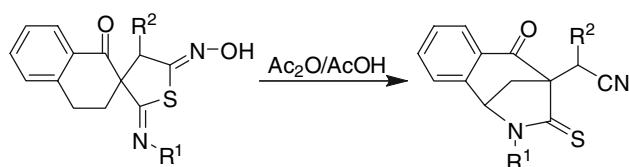
The  $\beta$ -ketothioamides are valuable building blocks for the synthesis of various nitrogen and sulfur heterocyclic compounds [1]. The versatility of nucleophilic centers of the  $\beta$ -ketothioamides facilitates their reactivity towards electrophilic reagents [2]. Acylating agents, e.g. oxalyl chloride [3], and alkylating agents, e.g. dihaloalkanes [4, 5], attack, mainly, the nitrogen and sulfur atoms affording thiazole or thiazine derivatives. In reactions with electrophilic olefins,  $\beta$ -ketothioamides have CH-acidic properties, e.g. reactions with  $\alpha,\beta$ -unsaturated aldehydes give 6-hydroxypiperidine-2-thiones [6]. We have recently found that conjugate addition of  $\beta$ -aminocycloalkenethioamides to maleimides and subsequent domino reaction provide easy access to polycyclic monothioimides [7]. However,  $\beta$ -ketothioamides also react with CH acids, e.g. malonitrile, yielding 6-aminopyridine-2-thiones [8]. In acidic medium some  $\gamma,\delta$ -unsaturated  $\beta$ -ketothioamides undergo intramolecular cyclization to 2,3-dihydro-4*H*-thiopyran-4-ones [9].  $\beta$ -Ketothioamides may react with some reagents which convert the thioamide group into five membered nitrogen heterocycles, e.g. the reaction with DMAD leads to pyrroles [10], and the reaction with TMS azide leads to 1,5-substituted tetrazoles [11].

Cyclic monothioimides have important biological properties. The effect of monothioimides on tumor necrosis factor alpha (TNF- $\alpha$ ) production by genetically modified B78H1 melanoma cells has been investigated [12]. It was found that monothioimide derivatives had significant activity as enhancers of TNF- $\alpha$  production. Monothioimides with spiro-annulated alicyclic rings have been successfully tested for anticonvulsant activity [13, 14]. Cyclic monothioimides have recently been used for

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Scheme 1

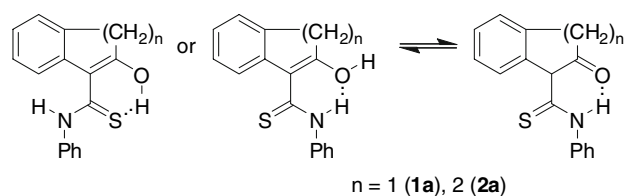
synthesis of protease-resistant and conformationally restricted peptidomimetics [15].

## Results and discussion

We have recently shown that *N*-aryl- $\beta$ -ketothioamides react readily with  $\beta$ -nitrostyrenes affording thiophene derivatives with high diastereoselectivity [16–19]. Because the thiophene skeletons contain oxime and arylimino groups they are sensitive under acidic conditions. The thiophenes derived from  $\alpha$ -unsubstituted  $\beta$ -ketothioamides treated with conc. HCl underwent a ring transformation to 5-thioxopyrrol-2-ones [16, 17, 19]. Thiophenes derived from 5 and 6-membered cyclic  $\beta$ -ketothioamides treated with conc. HCl underwent decomposition whereas they reacted smoothly under the acetylation conditions  $\text{Ac}_2\text{O}/\text{AcOH}$  yielding oxime esters [18]. We have found that thiophenes and their oxime esters spiro-annulated with a 1-tetralone skeleton underwent, under acidic conditions, an unique rearrangement giving polycyclic systems, e.g. 6-azabicyclo[3.2.1]octane [20] (Scheme 1).

Taking into account our results on rearrangement of compounds spiro-annulated with a 1-tetralone skeleton, in this work we studied the synthesis and the acidic rearrangements of thiophenes derived from some unreported thioamides containing 2-indanone and 2-tetralone ring systems.

The starting thioamides **1a–1c** and **2a–2c** were obtained, in moderate yields, by reaction of 2-indanone or 2-tetralone with sodium hydride and aryl isothiocyanates. The  $^1\text{H}$  NMR spectra of thioamides **1a–1c** in  $\text{CDCl}_3$  solutions contained singlets at  $\delta = 14.15$ – $14.10$  ppm, which indicated that the compounds exist almost exclusively in the enol form **A** in these solutions. The two alternative enol forms are shown in Scheme 2 [21]. Only traces of the keto form **B** were observed, e.g. the characteristic signals for the HC-1 protons at  $\delta \approx 4.65$  ppm. The ratio of the **A** and **B** forms in  $\text{CDCl}_3$  solutions, based on the integration in the  $^1\text{H}$  NMR spectra, was **A**:**B**  $\approx 40$ :1. The IR spectra of **1a–1c** in KBr pellets suggested the preference of the form **A** in the solid state also. In the  $^1\text{H}$  NMR spectra of the thioamides **2a–2c** in  $\text{CDCl}_3$  solutions two sets of signals for the **A** and **B** forms were clearly visible. The compounds existed predominantly in the enol form **A**. The ratio of

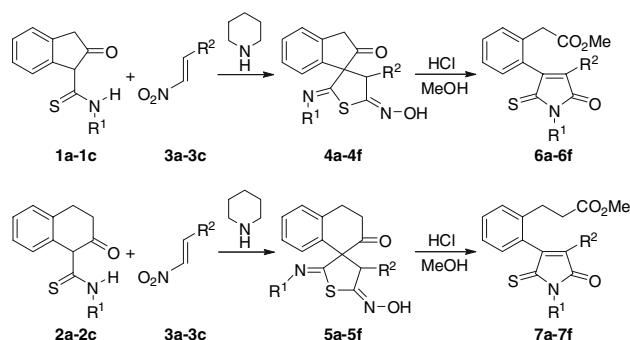


Scheme 2

forms **A** and **B** in  $\text{CDCl}_3$  solutions, **A**:**B**, ranged from 4.3:1 for **2c** to 4.9:1 for **2b**. In the  $^1\text{H}$  NMR spectra of **2a–2c** recorded in  $\text{DMSO}-d_6$  solutions an inverse ratio **A**:**B** has been observed, from 1:1.2 for **2c** to 1:1.7 for **2b**. The increase of amount of the keto form in more polar solvents is typical of  $\beta$ -ketothioamides [21]. The presence of a strong band at  $1,711\text{ cm}^{-1}$  in the IR spectrum of **2a** and a medium band at  $1,712\text{ cm}^{-1}$  in the IR spectrum of **2b** show the amount of the **B** form in the solid state. In the IR spectrum of **2c** the analogous C=O absorption band was not observed.

The thioamides **1a–1c** and **2a–2c** treated with (*E*)- $\beta$ -nitrostyrenes **3a–3c** in the presence of a catalytic amount of piperidine underwent a conjugated addition–cyclization to give the products **4a–4f** and **5a–5f** with thiophene rings spiro-annulated to the 2-indanone and 2-tetralone system (Scheme 3). The matched pairs of reagents had different aryl groups  $\text{R}^1$  and  $\text{R}^2$  (Table 1). The reactions of **1a–1c** with **3a–3c** were performed in acetonitrile, because we observed slow decomposition of the thioamides in the presence of primary alcohols (see below). However the desired products **4a–4f** were formed in moderate or poor yields (23–54%). The reactions of **2a–2c** with **3** were carried out in dry ethanol at boiling temperature, affording **5a–5f** in moderate to good yields (33–60%). The mechanism of the addition–cyclization has been reported in detail in our previous work [16–19].

The  $^1\text{H}$  NMR spectra of **4** and **5** contained the characteristic singlet of the oxime proton; e.g. in the  $^1\text{H}$  NMR spectrum of **4b** the oxime proton resonated at  $\delta = 11.88$  ppm and the analogous proton in the  $^1\text{H}$  NMR



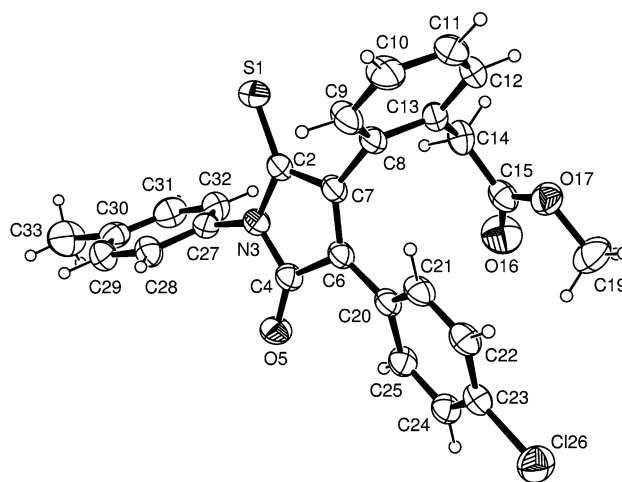
Scheme 3

**Table 1** Results from synthesis of the thiophene derivatives **4a–4f** and **5a–5f** and subsequent reactions with HCl in methanol (Scheme 3)

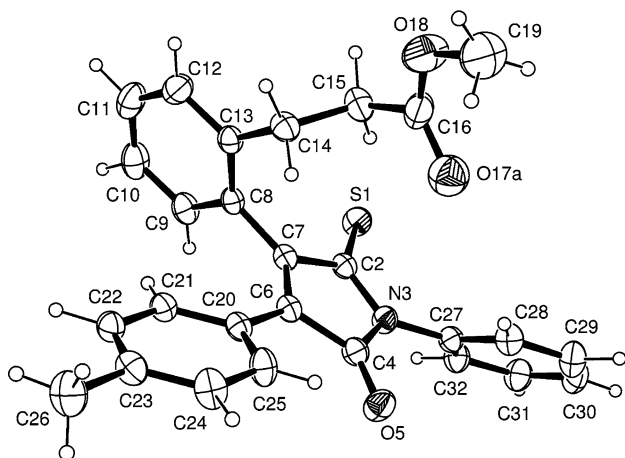
| Thioamide | Nitroalkene | R <sup>1</sup>                                   | R <sup>2</sup>                                   | Yield (%) | Yield (%) | Yield (%)    |
|-----------|-------------|--------------------------------------------------|--------------------------------------------------|-----------|-----------|--------------|
| <b>1a</b> | <b>3b</b>   | C <sub>6</sub> H <sub>5</sub>                    | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | <b>4a</b> | 36        | <b>6a</b> 26 |
| <b>1a</b> | <b>3c</b>   | C <sub>6</sub> H <sub>5</sub>                    | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | <b>4b</b> | 54        | <b>6b</b> 23 |
| <b>1b</b> | <b>3a</b>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | C <sub>6</sub> H <sub>5</sub>                    | <b>4c</b> | 46        | <b>6c</b> 11 |
| <b>1b</b> | <b>3c</b>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | <b>4d</b> | 56        | <b>6d</b> 20 |
| <b>1c</b> | <b>3a</b>   | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | C <sub>6</sub> H <sub>5</sub>                    | <b>4e</b> | 24        | <b>6e</b> 27 |
| <b>1c</b> | <b>3b</b>   | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | <b>4f</b> | 23        | <b>6f</b> 29 |
| <b>2a</b> | <b>3b</b>   | C <sub>6</sub> H <sub>5</sub>                    | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | <b>5a</b> | 46        | <b>7a</b> 41 |
| <b>2a</b> | <b>3c</b>   | C <sub>6</sub> H <sub>5</sub>                    | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | <b>5b</b> | 48        | <b>7b</b> 38 |
| <b>2b</b> | <b>3a</b>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | C <sub>6</sub> H <sub>5</sub>                    | <b>5c</b> | 47        | <b>7c</b> 32 |
| <b>2b</b> | <b>3c</b>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | <b>5d</b> | 33        | <b>7d</b> 22 |
| <b>2c</b> | <b>3a</b>   | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | C <sub>6</sub> H <sub>5</sub>                    | <b>5e</b> | 49        | <b>7e</b> 77 |
| <b>2c</b> | <b>3b</b>   | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | <b>5f</b> | 60        | <b>7f</b> 54 |

spectrum of **5b** resonated at  $\delta = 11.62$  ppm. The proton attached to C-4' of the spiro[indane-1,3'-thiophene] system of **4b** resonated at  $\delta = 5.00$  (s) ppm, and the protons attached to C-3 appeared as two doublets at  $\delta = 3.77$  and 3.63 ppm. The proton attached to C-4' of the spiro[naphthalene-1,3'-thiophene] system of **5b** resonated at  $\delta = 4.95$  (s) ppm, and the protons attached to C-3 and C-4 appeared as four signals coupled together (d  $\times$  d  $\times$  d) at  $\delta = 2.77$ , 2.62, 2.11, and 1.92 ppm.

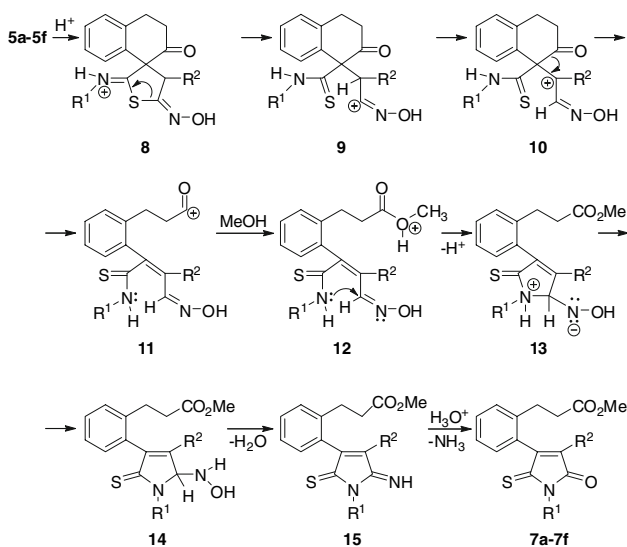
The reactivity of the thiophene derivatives **4a–4f** and **5a–5f**, which contain two exocyclic C=N bonds, towards acidic reagents has been investigated. The compounds were stable in acetic acid and acetic anhydride. When treated with hydrochloric acid in methanol solutions, however, **4a–4f** and **5a–5f** underwent a rearrangement involving transformation of the thiophene ring and breaking of the alicyclic ring in the proximity of the carbonyl group. The reaction mixture after work up afforded the products **6a–6f** in moderate to poor yields (11–29%) and **7a–7f** in moderate to good yields (22–77%). The IR spectra of the products contained the characteristic bands indicating ester functionality, e.g. broad bands at  $1,733\text{ cm}^{-1}$  (C=O) and at  $1,150\text{ cm}^{-1}$  (C–O) in the IR spectrum of **6b**. In the <sup>1</sup>H NMR spectrum of **6b** the ester methyl group appeared as singlet at  $\delta = 3.45$  ppm, overlapping the two doublets of the AB system of the methylene fragment adjacent to the ester group. The HSQC spectrum revealed correlations of the protons with two carbon atoms: the methyl carbon at  $\delta = 52.0$  and the methylene carbon at  $\delta = 39.6$  ppm. The <sup>13</sup>C NMR spectrum of **6b** also revealed the thiocarbonyl group at  $\delta = 199.3$  ppm. Other signals, especially those of the aromatic and alkene fragments appear very close to each other and cannot be assigned. Therefore, the structure of compounds **6** and **7** was unequivocally determined on the basis of crystallographic measurements.

**Fig. 1** A perspective view of the molecule of **6d** with the crystallographic atom numbering scheme

Two crystalline compounds **6d** and **7a** were chosen and submitted to X-ray crystal analysis. Perspective views of the molecules of **6d** and **7a** with the crystallographic atom numbering are shown in Figs. 1 and 2, respectively. Despite the similar structures, there is one rather significant difference. The terminal COOCH<sub>3</sub> group has one more single bond separating it from the phenyl ring and thus more rotational freedom. This manifests itself in the disorder of this substituent in **7a**, whereas the same moiety does not show any disorder in **6d**. This could be modeled in part by splitting the atom O17 in **7a**, with both positions rotated by about 37 degrees. Also, the atomic displacement parameters of the remaining atoms in this moiety are somewhat high. As a result there is a residual electron density peak of  $0.63\text{ e}\text{\AA}^{-3}$  in the final Fourier map of **7a**. The conformations of the molecules of both **6d** and **7a** are stabilized by an intramolecular close contact C25–



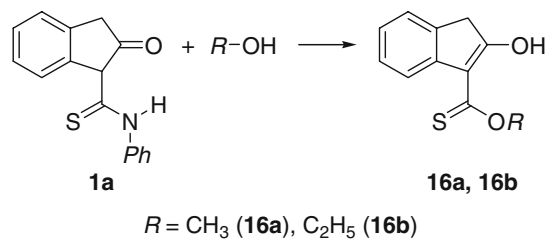
**Fig. 2** A perspective view of the molecule of **7a** with the crystallographic atom numbering scheme. The oxygen atom O17 has two disordered positions, of which only one is presented in the figure for clarity



**Scheme 4**

H25...O5 (in both cases). Neither crystal structure shows typical intermolecular hydrogen bonds.

The reaction of **4a–4f** and **5a–5f** with HCl and MeOH leading to the formation of **6a–6f** and **7a–7f** might be explained by a mechanism shown in Scheme 4 for **5a–5f**. Initial protonation of the arylimino group leads to opening of the thiophene ring and formation of hydroxyiminium ion **9**. The key step of rearrangement is the conversion of the ion **9** to the carbocation **10** by a 1,2-hydrogen shift. The tertiary carbocation **10**, which is also stabilized by the aryl group  $R^2$ , undergoes subsequent rearrangement involving the formation of a C=C double bond and the opening of the tetralone ring at a position adjacent to the carbonyl group. The newly formed acylium ion **11** reacts with the solvent



**Scheme 5**

MeOH affording the methyl ester. Nucleophilic attack of the thioamide nitrogen on the carbon of the aldoxime group leads to the closure of a pyrrole ring. Subsequent elimination of a water molecule from **14** followed by a spontaneous hydrolysis of the imine **15** provides the products **7a–7f**.

We observed unusual reactivity of the thioamides derived from 2-indanone **1a–1c** towards primary alcohols. For example, the thioamide **1a** reacted with boiling methanol or ethanol affording appropriate thiocarboxylic acid *O*-esters **16a** or **16b** in good yields of 43 and 61% (Scheme 5). The direction of these reactions is opposite to the typically observed synthesis of amides from esters [22, 23]. The new reactions seem to be the only method of transformation of thioamides into *O*-thioesters, because the reactions of thioamides with alkylating agents, e.g. alkyl iodides or dimethyl sulfate, give *S*-thioesters [24, 25]. In the  $^{13}\text{C}$  NMR spectrum of **16b** in  $\text{CDCl}_3$  solution the thiocarbonyl carbon atom appeared at  $\delta = 204.6$  ppm. The  $^1\text{H}$  NMR spectrum of **16b** contained a triplet at  $\delta = 1.64$  ppm ( $\text{CH}_3$ ) and quartet at  $\delta = 4.74$  ppm ( $\text{CH}_2$ ) for the ethyl group, which clearly indicate the presence of the *O*-ester group. The singlet at  $\delta = 13.84$  ppm assigned to the OH proton shows that the product **16b** exists almost exclusively in enol form in  $\text{CDCl}_3$  solution.

In conclusion, we have developed new applications of tandem conjugate addition–cyclization reactions for synthesis of functionalized thiophenes and pyrroles. This study revealed a unique rearrangement of spiro-annulated thiophene derivatives under acidic conditions involving breakage of a C–C single bond and ring transformations of iminothiophenes to pyrroles. The results presented provide easy access to new heterocycles with potential biological properties.

## Experimental

Melting points were determined on a Boetius hot-stage apparatus and are corrected. IR spectra were run in KBr pellets on a Bruker IFS 48 spectrometer. Mass spectra were obtained on a Finnigan Mat 95 (EI) or a Bruker Esquire 3000 (ESI) mass spectrometer. NMR spectra were recorded

on a Bruker AMX 500 spectrometer ( $^1\text{H}$ , 500.14 MHz;  $^{13}\text{C}$ , 125.76 MHz) or Bruker Avance II 300 MHz in  $\text{CDCl}_3$  or  $\text{DMSO}-d_6$  using TMS as internal standard. Microanalyses (C, H, N) of the crystalline compounds were performed with a Euro EA 3000 Elemental Analyzer; the results agreed satisfactorily ( $\pm 0.3\%$ ) with calculated values. Diffraction data were collected using a Nonius Kappa CCD diffractometer with graphite monochromated  $\text{Mo}-K_\alpha$  radiation.

#### General procedure for synthesis of thioamides **1a–1c**, **2a–2c**

To a vigorously stirred suspension of sodium hydride (50 mmol) in  $90\text{ cm}^3$  dry DMF at  $-10\text{ }^\circ\text{C}$ , 2-indanone (for **1a–1c**; 6.60 g, 50 mmol) or 2-tetralone (for **2a–2c**; 7.30 g, 50 mmol) was added slowly, over a period of 1 h, so the temperature did not exceed  $0\text{ }^\circ\text{C}$ . After the gas was evolved, a solution of the appropriate aryl isothiocyanate (50 mmol) in  $10\text{ cm}^3$  dry DMF was added dropwise. The mixture was stirred for 3 h at  $-10\text{ }^\circ\text{C}$ , left in a fridge overnight, then added slowly to  $50\text{ cm}^3$  1 M HCl, and acidified with 2 M HCl. After 3 h most of the water–DMF solution was decanted. The oily residue was dissolved in  $\text{CH}_2\text{Cl}_2$ , washed twice with 1 M HCl and water, then evaporated and purified by flash chromatography on silica gel using  $\text{CH}_2\text{Cl}_2$  as eluent. The crude **2a–2c** was treated with  $\text{Et}_2\text{O}$ , cooled in a fridge, and collected by filtration. The crude **1a–1c** was dissolved in MeCN at  $50\text{ }^\circ\text{C}$  and treated with  $50\text{ cm}^3$  2 M HCl for hydrolysis of imine side-products. After evaporation of the solvent the product was purified again by flash chromatography on silica gel, using  $\text{CH}_2\text{Cl}_2$  as eluent, and crystallized from  $\text{Et}_2\text{O}$ .

#### 2,3-Dihydro-2-oxo-N-phenyl-1H-indene-1-carbothioamide (**1a**, $\text{C}_{16}\text{H}_{13}\text{NOS}$ )

Orange crystals (7.41 g, 56%);  $R_f = 0.41$  ( $\text{CHCl}_3$ ); m.p.:  $127\text{--}128\text{ }^\circ\text{C}$ ;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ): 97% of enol **A** + 3% of ketone **B**,  $\delta = 14.15$  (s, 1H, OH **A**), 9.29 (bs, 1H, NH **B**), 8.87 (bs, 1H, NH **A**), 7.53 (d, 2H,  $^3J = 8.0$  Hz, CH arom **B**), 7.57 (d, 2H,  $^3J = 8.0$  Hz, CH arom **A**), 7.50–7.27 (m, 6CH arom **A** + 6CH arom **B**), 7.17 (ddd,  $^3J \approx 7.4, 7.4, ^4J = 1.0$  Hz, 1CH arom **A** + 1CH arom **B**), 4.66 (s, 1H, HC-1 **B**), 3.67 (s, 2H **A** + 2H **B**,  $\text{H}_2\text{C}-3$  **A** + **B**) ppm;  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 188.8, 182.1, 140.2, 137.2, 132.7, 129.1, 127.30, 127.27, 125.1, 124.8, 124.2, 117.7, 111.4, 39.2$  ppm; IR (KBr):  $\bar{\nu} = 3,333, 3,041, 3,003, 2,923, 1,549, 1,515, 1,494, 1,472, 1,457, 1,417, 1,359, 1,314, 1,291, 1,227, 1,139, 913\text{ cm}^{-1}$ ; MS (70 eV):  $m/z$  (%) = 267.1 (51)  $[\text{M}]^+$ , 233.1 (18), 204.1 (20), 174.0 (25), 146.0 (35), 132.1 (17), 131.1 (18), 102.0 (20), 93.0 (100), 77.0 (17).

#### 2,3-Dihydro-N-(4-methylphenyl)-2-oxo-1H-indene-1-carbothioamide (**1b**, $\text{C}_{17}\text{H}_{15}\text{NOS}$ )

Dark orange crystals (9.89 g, 70%); m.p.:  $121\text{--}122\text{ }^\circ\text{C}$ ;  $R_f = 0.42$  ( $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ): 97% of enol **A** + 3% of ketone **B**,  $\delta = 14.15$  (s, 1H, OH **A**), 9.21 (bs, 1H, NH **B**), 8.87 (bs, 1H, NH **A**), 7.53 (d, 2H,  $^3J = 8.4$  Hz, CH arom **B**), 7.46–7.36 (m, 4CH arom **A** + 2CH arom **B**), 7.34–7.22 (m, 3CH arom **A** + 3CH arom **B**), 7.16 (ddd,  $^3J \approx 7.5, 7.5, ^4J = 0.9$  Hz, 1CH arom **A** + 1CH arom **B**), 4.65 (s, 1H, HC-1 **B**), 3.66 (s, 2H **A** + 2H **B**,  $\text{H}_2\text{C}-3$  **A** + **B**), 2.39 (s, 3H,  $\text{CH}_3$  **A**), 2.34 (s, 3H,  $\text{CH}_3$  **B**) ppm;  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 188.8, 181.8, 140.2, 137.3, 134.6, 132.7, 129.7, 127.3, 125.1, 124.7, 124.1, 117.6, 111.3, 39.1, 21.2$  ppm; IR (KBr):  $\bar{\nu} = 3,365, 3,314, 3,079\text{--}3,018, 2,951, 1,562, 1,512, 1,488, 1,457, 1,401, 1,377, 1,299, 1,212, 1,142, 906\text{ cm}^{-1}$ ; MS (70 eV):  $m/z$  (%) = 281.1 (31)  $[\text{M}]^+$ , 247.1 (18), 174.0 (29), 146.0 (36), 145.0 (19), 132.1 (10), 107.1 (100), 106.1 (64), 102.1 (22), 77.0 (12).

#### N-(4-Chlorophenyl)-2,3-dihydro-2-oxo-1H-indene-1-carbothioamide (**1c**, $\text{C}_{16}\text{H}_{12}\text{ClNOS}$ )

Orange crystals (9.516 g, 63%); m.p.:  $151\text{--}153\text{ }^\circ\text{C}$ ;  $R_f = 0.44$  ( $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ): 97% of enol **A** + 3% of ketone **B**,  $\delta = 14.10$  (s, 1H, OH **A**), 9.31 (bs, 1H, NH **B**), 8.83 (bs, 1H, NH **A**), 7.66 (d, 2H,  $^3J = 8.7$  Hz, CH arom **B**), 7.51 (d, 2H,  $^3J = 8.7$  Hz, CH arom **A**), 7.44–7.37 (m, 4CH arom **A** + 4CH arom **B**), 7.36–7.28 (dd,  $^3J \approx 7.4, 7.4$  Hz, 1CH arom **A** + 1CH arom **B**), 7.18 (ddd,  $^3J \approx 7.4, 7.4, ^4J = 1.1$  Hz, 1CH arom **A** + 1CH arom **B**), 4.64 (s, 1H, HC-1 **B**), 3.67 (s, 2H,  $\text{H}_2\text{C}-3$  **A** + **B**) ppm;  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 188.9, 182.5, 140.0, 135.7, 132.7, 132.6, 129.3, 127.3, 126.4, 124.8, 124.3, 117.6, 111.4, 39.2$  ppm; IR (KBr):  $\bar{\nu} = 3,369, 3,090\text{--}3,021, 2,925\text{--}2,851, 1,548, 1,502, 1,465, 1,430, 1,392, 1,378, 1,273, 1,138, 1,103, 1,090, 901\text{ cm}^{-1}$ ; MS (70 eV):  $m/z$  (%) = 303.0 (8)  $[\text{M} + 2]^+$ , 301.1 (51)  $[\text{M}]^+$ , 204.1 (12), 174.0 (41), 146.0 (51), 145.0 (24), 132.1 (10), 129.0 (31), 127.0 (100), 102.1 (26), 65.1 (11).

#### 1,2,3,4-Tetrahydro-2-oxo-N-phenylnaphthalene-1-carbothioamide (**2a**, $\text{C}_{17}\text{H}_{15}\text{NOS}$ )

Yellow crystals (12.65 g, 90%); m.p.:  $129\text{--}131\text{ }^\circ\text{C}$ ;  $R_f = 0.39$  ( $\text{CHCl}_3$ );  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ): 83% of enol **A** + 17% of ketone **B**,  $\delta = 14.01$  (bs, 1H, OH **A**), 9.29 (bs, 1H, NH **B**), 8.87 (bs, 1H, NH **A**), 7.67 (d, 2H,  $^3J = 7.7$  Hz, CH arom **B**), 7.56–7.08 (m, 9CH arom **A** + 7CH arom **B**), 4.79 (s, 1H, C-1 **B**), 3.34 (ddd, 1H,  $^2J = 15.7$  Hz,  $^3J = 9.7, 5.8$  Hz, HC-3 **B**), 3.01 (ddd, 1H,  $^2J = 15.7$  Hz,  $^3J \approx 6, 5.5$  Hz, HC-3 **B**), 2.91 (dt, 1H,  $^2J = 17.5$  Hz,  $^3J \approx 5.5$  Hz, HC-4 **B**), 2.82 (t, 2H,  $^3J \approx 7$  Hz,  $\text{H}_2\text{C}-3$  **A**), 2.63–2.52 (m,  $^3J \approx 7$  Hz,  $\text{H}_2\text{C}-4$

**A**, HC-4 **B**) ppm;  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ ):  $\delta = 206.7, 190.9, 174.0, 165.7, 138.4, 137.6, 135.7, 133.7, 132.8, 129.4, 129.0, 128.9, 128.5, 128.2, 127.4, 127.11, 127.07, 126.98, 125.7, 124.9, 124.8, 123.4, 109.7, 69.5, 37.2, 31.1, 27.9, 27.7$  ppm; IR (KBr):  $\bar{\nu} = 3,291$  (N–H), 3,129, 3,060, 3,034, 2,968–2,850, 1,711 (C=O), 1,596, 1,541, 1,493, 1,412, 1,231, 1,205, 1,158, 1,117  $\text{cm}^{-1}$ ; MS (70 eV):  $m/z$  (%) = 281.1 (96)  $[\text{M}]^+$ , 248.1 (27), 188 (49), 160.0 (23), 115.0 (41), 93 (100), 77.0 (25).

*1,2,3,4-Tetrahydro-N-(4-methylphenyl)-2-oxonaphthalene-1-carbothioamide (2b, C<sub>18</sub>H<sub>17</sub>NOS)*

Yellow crystals (11.59 g, 79%); m.p.: 113–115 °C;  $R_f = 0.40$  ( $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ): 82% of enol **A** + 18% of ketone **B**,  $\delta = 13.99$  (bs, 1H, OH **A**), 9.22 (bs, 1H, NH **B**), 8.83 (bs, 1H, NH **A**), 7.50 (d, 2H,  $^3J = 8.4$  Hz, CH arom **B**), 7.44–7.06 (m, 8CH arom **A** + 6CH arom **B**), 4.77 (s, 1H, C-1 **B**), 3.34 (ddd, 1H,  $^2J = 15.5$  Hz,  $^3J = 9.4, 5.8$  Hz, HC-3 **B**), 3.04–2.86 (m, 2H, HC-3 **B**, HC-4 **B**), 2.81 (t, 2H,  $^3J \approx 7$  Hz, H<sub>2</sub>C-3 **A**), 2.64–2.50 (m, 2H **A** + 1H **B**,  $^3J \approx 7$  Hz, H<sub>2</sub>C-4 **A**, HC-4 **B**), 2.36 (s, 3H, CH<sub>3</sub> **A**), 2.33 (s, 3H, CH<sub>3</sub> **B**) ppm;  $^1\text{H}$  NMR (300 MHz, DMSO-*d*<sub>6</sub>): 37% of enol **A** + 63% of ketone **B**,  $\delta = 12.01$  (bs, 1H, NH **B**), 11.43 (bs, 1H, OH **A**), 10.32 (bs, 1H, NH **A**), 7.73 (bd, 2H,  $^3J \approx 7.4$  Hz, CH arom **A**), 7.65 (d, 2H,  $^3J = 8.5$  Hz, CH arom **B**), 7.33–6.90 (m, 6CH arom **A** + 6CH arom **B**), 5.02 (s, 1H, HC-1 **B**), 3.47 (ddd, 1H,  $^2J = 15.6$  Hz,  $^3J = 9.4, 5.6$  Hz, HC-3 **B**), 2.99 (dt, 1H,  $^2J = 15.6$  Hz,  $^3J \approx 6$  Hz, HC-3 **B**), 2.88–2.72 (m,  $^2J = 16.3$  Hz,  $^3J \approx 6$  Hz, HC-4 **B**, H<sub>2</sub>C-3 **A**), 2.61 (ddd, 1H,  $^2J = 16.3$  Hz,  $^3J = 9.4, 5.8$  Hz, HC-4 **B**), 2.48–2.40 (m, 2H, H<sub>2</sub>C-4 **A**), 2.31 (s, 3H **A** + 3H **B**, CH<sub>3</sub> **A** + **B**) ppm;  $^{13}\text{C}$  NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 206.2, 198.1, 194.0, 155.8, 137.0, 136.7, 136.4, 135.8, 135.7, 134.8, 131.9, 128.9, 128.7, 128.1, 127.7, 126.9, 126.8, 126.5, 126.2, 124.0, 123.4, 123.2, 123.0, 114.4, 66.7, 38.1, 28.2, 27.4, 27.3, 20.6$  ppm; IR (KBr):  $\bar{\nu} = 3,322, 3,211, 3,059, 3,032, 2,956$ –2,832, 1,712 (C=O), 1,597, 1,560, 1,513, 1,483, 1,431, 1,412, 1,312, 1,292, 1,202, 1,188, 1,142, 921  $\text{cm}^{-1}$ ; ESI-MS:  $m/z$  (%) = 318.0 (31)  $[\text{M} + \text{Na}]^+$ , 296.1 (100)  $[\text{M} + \text{H}]^+$ .

*N-(4-Chlorophenyl)-1,2,3,4-tetrahydro-2-oxonaphthalene-1-carbothioamide (2c, C<sub>17</sub>H<sub>14</sub>ClNOS)*

Yellow crystals (13.87 g, 88%); m.p.: 99–102 °C;  $R_f = 0.42$  ( $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ): 81% of enol **A** + 19% of ketone **B**,  $\delta = 13.98$  (bs, 1H, OH **A**), 9.34 (bs, 1H, NH **B**), 8.81 (bs, 1H, NH **A**), 7.65 (d, 2H,  $^3J = 8.8$  Hz, CH arom **B**), 7.47 (d, 2H,  $^3J = 8.8$  Hz, CH arom **A**), 7.40–7.08 (m, 6CH arom **A** + 6CH arom **B**), 4.78 (s, 1H, HC-1 **B**), 3.32 (ddd, 1H,  $^2J = 15.6$  Hz,  $^3J = 9.7, 6.0$  Hz, HC-3 **B**), 3.01 (ddd, 1H,  $^2J = 15.6$  Hz,  $^3J \approx 6.0, 5.2$  Hz, HC-3 **B**), 2.91–2.75 (t, 2H **A**,

$^3J \approx 7$  Hz, H<sub>2</sub>C-3 **A** + m, 1H **B**, HC-4 **B**), 2.63–2.52 (m, 2H **A**,  $^3J \approx 7$  Hz, H<sub>2</sub>C-4 **A**, + 1H **B**, HC-4 **B**) ppm;  $^1\text{H}$  NMR (300 MHz, DMSO-*d*<sub>6</sub>): 45% of enol **A** + 55% of ketone **B**,  $\delta = 12.15$  (bs, 1H, NH **B**), 11.62 (bs, 1H, OH **A**), 10.18 (bs, 1H, NH **A**), 7.94 (bd, 2H, CH arom **A**), 7.85 (d, 2H,  $^3J = 8.8$  Hz, CH arom **B**), 7.48 (m,  $^3J = 8.8$  Hz, 2CH arom **B** + 2CH arom **A**), 7.32–7.20 (m, 4CH arom **B**), 7.15–6.94 (m, 4CH arom **A**), 5.04 (s, 1H, HC-1 **B**), 3.44 (ddd, 1H,  $^2J = 15.5$  Hz,  $^3J = 9.3, 5.4$  Hz, HC-3 **B**), 3.00 (dt, 1H,  $^2J = 15.5$  Hz,  $^3J \approx 6$  Hz, HC-3 **B**), 2.89–2.71 (m, 2H **A**, H<sub>2</sub>C-3 **A** + 1H **B**,  $^2J = 16.3$  Hz,  $^3J \approx 6$  Hz, HC-4 **B**), 2.62 (ddd, 1H,  $^2J = 16.3$  Hz,  $^3J = 9.3, 5.9$  Hz, HC-4 **B**), 2.48–2.40 (m, 2H, H<sub>2</sub>C-4 **A**) ppm;  $^{13}\text{C}$  NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 207.2, 200.1, 156.3, 139.7, 139.0, 137.9, 136.9, 135.9, 132.9, 131.3, 130.7, 129.7, 129.5, 129.3, 129.0, 128.2, 128.1, 127.8, 127.4, 126.1, 125.2, 124.1, 68.2, 39.3, 29.3, 28.6, 28.5$  ppm; IR (KBr):  $\bar{\nu} = 3,309, 3,089$ –3,026, 2,942–2,817, 1,584, 1,561, 1,508, 1,483, 1,413, 1,393, 1,311, 1,285, 1,210, 1,194, 1,139, 1,091  $\text{cm}^{-1}$ ; ESI-MS:  $m/z$  (%) = 318.2 (44), 316.3 (100)  $[\text{M} + \text{H}]^+$ .

*General procedure for synthesis of spiro[indane-1,3'-thiophenes] 4a–4f*

To a solution of the appropriate thioanilide **1a–1c** (5 mmol) in 20 cm<sup>3</sup> MeCN (for **1c** 60 cm<sup>3</sup>) two drops of piperidine were added at 50 °C. The appropriate (*E*)- $\beta$ -nitrostyrene **3a–3c** (5.5 mmol) was then added slowly over a period of 0.5 h. The mixture was stirred for 0.5 h at 50 °C, then concentrated in vacuo to half of the initial volume, and cooled in a fridge. The product was isolated by filtration and recrystallized from EtOAc.

*2,3,4',5'-Tetrahydro-5'-(hydroxyimino)-4'-(4-methylphenyl)-2'-(phenylimino)spiro[1H-indene-1,3'-(2'H)-thiophene]-2-one (4a, C<sub>25</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (734 mg, 36%); m.p.: 226–228 °C;  $^1\text{H}$  NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 11.66$  (s, 1H, NOH), 7.81 (d, 1H,  $^3J = 7.5$  Hz, CH arom), 7.49–7.29 (m, 4H, CH arom), 7.22–7.13 (m, 2H, CH arom), 6.97 (d, 2H,  $^3J = 8.0$  Hz, CH arom), 6.88–6.77 (m, 4H, CH arom), 5.02 (s, 1H, HC-4'), 3.56 (d, 1H,  $^2J = 23.1$  Hz, HC-3), 2.67 (d, 1H,  $^2J = 23.1$  Hz, HC-3), 2.19 (s, 3H, CH<sub>3</sub>) ppm;  $^{13}\text{C}$  NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 210.8, 167.1, 151.7, 150.3, 139.6, 137.3, 136.8, 130.0, 129.8, 129.3, 128.5, 128.3, 127.7, 125.9, 125.0, 124.1, 119.0, 73.8, 56.9, 42.4, 20.5$  ppm; IR (KBr):  $\bar{\nu} = 3,250, 3,094$ –3,011, 2,920–2,858, 1,751, 1,655, 1,633, 1,592, 1,485, 1,428, 1,134, 1,118, 1,071, 964  $\text{cm}^{-1}$ ; MS (70 eV):  $m/z$  (%) = 411.8 (100)  $[\text{M}]^+$ , 277.1 (13), 248.1 (14), 234.1 (22), 233.0 (38), 232.0 (16), 205.1 (16), 204.1 (32), 77.2 (11).



*4'-(4-Chlorophenyl)-2,3,4',5'-tetrahydro-5'-(hydroxyimino)-2'-(phenylimino)spiro[1H-indene-1,3'(2'H)-thiophene]-2-one (4b, C<sub>24</sub>H<sub>17</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (1.164 g, 54%); m.p.: 185–187 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>): δ = 11.88 (s, 1H, NOH), 7.44–7.15 (m, 8H, CH arom), 7.07 (d, 2H, <sup>3</sup>J = 8.5 Hz, CH arom), 6.88 (d, 1H, <sup>3</sup>J = 7.6 Hz, CH arom), 6.79 (d, 2H, <sup>3</sup>J = 7.3 Hz, CH arom), 5.00 (s, 1H, HC-4'), 3.77 (d, 1H, <sup>2</sup>J = 23.1 Hz, HC-3), 3.63 (d, 1H, <sup>2</sup>J = 23.1 Hz, HC-3) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>): δ = 210.7, 167.3, 151.4, 149.9, 137.9, 137.0, 134.0, 132.3, 131.0, 129.4, 128.7, 128.0, 127.0, 125.5, 125.3, 124.9, 119.2, 72.8, 53.6, 42.1 ppm; IR (KBr):  $\bar{\nu}$  = 3,271, 3,054, 2,956–2,853, 1,754, 1,593, 1,556, 1,513, 1,494, 1,440, 1,418, 1,375, 1,183, 1,130, 1,094, 1,061, 955 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 432.0 (6) [M]<sup>+</sup>, 429.3 (36), 428.3 (100), 427.2 (28), 413.1 (46), 406.2 (28), 405.2 (30), 404.2 (79), 335.0 (21), 267.1 (46), 234.1 (76), 233.1 (56), 205.1 (39), 204.1 (61), 169.0 (27), 146.0 (24), 136.0 (35), 102.0 (33), 93.0 (41), 77.0 (23).

*2,3,4',5'-Tetrahydro-5'-(hydroxyimino)-2'-[(4-methylphenyl)imino]-4'-phenylspiro[1H-indene-1,3'(2'H)-thiophene]-2-one (4c, C<sub>25</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (941 mg, 46%); m.p.: 199–201 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>): δ = 11.76 (s, 1H, NOH), 7.34–7.12 (m, 8H, CH arom), 7.09–7.02 (m, 2H, CH arom), 6.82 (d, 1H, <sup>3</sup>J = 7.6 Hz, CH arom), 6.70 (d, 2H, <sup>3</sup>J = 8.3 Hz, CH arom), 4.90 (s, 1H, HC-4'), 3.75 (d, 1H, <sup>2</sup>J = 23.0 Hz, HC-3), 3.59 (d, 1H, <sup>2</sup>J = 23.0 Hz, HC-3), 2.29 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>): δ = 210.8, 167.0, 152.0, 147.5, 138.2, 137.0, 135.0, 134.5, 129.8, 129.2, 128.4, 127.9, 127.6, 126.8, 125.6, 124.7, 119.2, 72.9, 54.4, 42.1, 20.4 ppm; IR (KBr):  $\bar{\nu}$  = 3,284, 3,065–3,002, 2,951, 2,951, 1,757, 1,554, 1,514, 1,416, 1,380, 1,308, 1,286, 1,185, 1,133, 1,061 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 412.1 (100) [M]<sup>+</sup>, 281.1 (10), 263.1 (14), 247.1 (35), 246.1 (16), 234.1 (12), 233.1 (14), 220.1 (17), 218.1 (27), 205.1 (15), 204.1 (16), 191.1 (18), 174.0 (11), 146.1 (16), 107.1 (48), 106.1 (28), 102.0 (18), 91.0 (19), 77.0 (13).

*4'-(4-Chlorophenyl)-2,3,4',5'-tetrahydro-5'-(hydroxyimino)-2'-[(4-methylphenyl)imino]spiro[1H-indene-1,3'(2'H)-thiophene]-2-one (4d, C<sub>25</sub>H<sub>19</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (1.256 g, 56%); m.p.: 184–186 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>): δ = 11.84 (s, 1H, NOH), 7.34–7.16 (m, 7H, CH arom), 7.07 (d, 2H, <sup>3</sup>J = 8.5 Hz, CH arom), 6.90 (d, 1H, <sup>3</sup>J = 7.5 Hz, CH arom), 6.70 (d, 2H, <sup>3</sup>J = 8.2 Hz, CH arom), 4.97 (s, 1H, HC-4'), 3.76 (d, 1H, <sup>2</sup>J = 23.1 Hz, HC-3), 3.61 (d, 1H, <sup>2</sup>J = 23.1 Hz, HC-3), 2.28 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>): δ = 210.8, 167.1, 151.7, 150.3, 139.6, 137.3, 136.8,

130.0, 129.8, 129.3, 128.5, 128.3, 127.7, 125.9, 125.0, 124.1, 119.0, 73.8, 56.9, 42.4, 20.5 ppm; IR (KBr):  $\bar{\nu}$  = 3,260, 3,100–3,036, 2,961, 2,916, 1,753, 1,592, 1,557, 1,517, 1,491, 1,418, 1,377, 1,183, 1,131, 1,091, 1,063 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 446.1 (2) [M]<sup>+</sup>, 420.1 (36), 419.1 (37), 418.1 (93), 417.1 (32), 281.1 (36), 280.1 (22), 269.1 (28), 268.1 (22), 265.1 (27), 254.1 (13), 250.1 (13), 249.1 (19), 248.1 (100), 247.1 (37), 236.1 (17), 219.1 (22), 218.1 (22), 205.1 (19), 204.1 (32), 203.1 (14), 150.0 (28), 146.0 (17), 138.0 (22), 132.1 (14), 125.0 (24), 107.1 (31), 65.1 (26), 63.0 (21).

*2'-[(4-Chlorophenyl)imino]-2,3,4',5'-tetrahydro-5'-(hydroxyimino)-4'-phenylspiro[1H-indene-1,3'(2'H)-thiophene]-2-one (4e, C<sub>24</sub>H<sub>17</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (522 mg, 24%); m.p.: 211–214 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>): δ = 11.84 (s, 1H, NOH), 7.44 (d, 2H, <sup>3</sup>J = 8.7 Hz, CH arom), 7.32–7.10 (m, 6H, CH arom), 7.09–7.02 (m, 2H, CH arom), 6.83 (d, 2H, <sup>3</sup>J = 8.7 Hz, CH arom), 6.76 (d, 1H, <sup>3</sup>J = 7.7 Hz, CH arom), 4.93 (s, 1H, HC-4'), 3.80–3.59 (m, 2H, <sup>2</sup>J = 23.0 Hz, H<sub>2</sub>C-3) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>): δ = 210.5, 168.8, 151.6, 148.8, 138.0, 137.0, 135.1, 129.41, 129.37, 129.1, 128.5, 128.0, 127.7, 126.8, 125.7, 124.7, 121.1, 73.0, 54.5, 42.0 ppm; IR (KBr):  $\bar{\nu}$  = 3,205, 3,067, 3,037, 2,921–2,830, 1,756, 1,647, 1,616, 1,484, 1,457, 1,422, 1,390, 1,209, 1,146, 1,123, 965 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 434.1 (40) [M + 2]<sup>+</sup>, 432.1 (100) [M]<sup>+</sup>, 387.1 (11), 267.1 (14), 263.1 (16), 246.1 (11), 234.1 (11), 233.1 (15), 220.1 (19), 219.1 (12), 218.1 (15), 217.1 (12), 215.1 (13), 205.1 (16), 204.1 (24), 191.1 (20).

*2'-[(4-Chlorophenyl)imino]-2,3,4',5'-tetrahydro-5'-(hydroxyimino)-4'-(4-methylphenyl)spiro[1H-indene-1,3'(2'H)-thiophene]-2-one (4f, C<sub>25</sub>H<sub>19</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (527 mg, 23%); m.p.: 191–193 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>): δ = 11.81 (s, 1H, NOH), 7.43 (d, 2H, <sup>3</sup>J = 8.7 Hz, CH arom), 7.33–7.24 (m, 2H, CH arom), 7.17 (dd, <sup>3</sup>J ≈ 7.4, <sup>4</sup>J = 1.9 Hz, CH arom), 6.99 (d, 2H, <sup>3</sup>J = 8.2 Hz, CH arom), 6.92 (d, 2H, <sup>3</sup>J = 8.2 Hz, CH arom), 6.87–6.78 (m, 3H, CH arom), 4.88 (s, 1H, HC-4'), 3.79–3.55 (m, 2H, <sup>2</sup>J = 23.1 Hz, H<sub>2</sub>C-3), 2.20 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>): δ = 210.6, 168.9, 151.6, 148.7, 138.1, 137.0, 136.8, 131.8, 129.40, 129.36, 129.0, 128.5, 126.9, 125.6, 124.8, 121.1, 73.1, 54.3, 42.1, 20.5 ppm; IR (KBr):  $\bar{\nu}$  = 3,267, 3,095–3,021, 2,961, 2,920, 1,754, 1,555, 1,513, 1,417, 1,379, 1,283, 1,186, 1,131, 1,091, 1,065, 1,015 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 446.1 (3) [M]<sup>+</sup>, 420.1 (35), 419.1 (35), 418.1 (84), 417.1 (34), 301.0 (17), 270.1 (38), 269.1 (27), 268.1 (100), 267.1 (30), 265.1 (56), 248.1 (28), 236.1 (23), 205.1 (26), 204.1 (37), 174.0 (35), 169.0 (22), 146.0 (42), 129.0 (26), 127.0 (65), 118.1 (41), 117.1 (22), 115.1 (36), 105.1 (36), 102.1 (32), 91.0 (24).

*General procedure for synthesis of spiro[naphthalene-1,3'-thiophenes] 5a–5f*

To a solution of the appropriate thioanilide **2a–2c** (5 mmol) and (*E*)- $\beta$ -nitrostyrene **3a–3c** (5 mmol) in 50 cm<sup>3</sup> dry EtOH, two drops of piperidine were added. The mixture was gently heated under reflux for 3 h then the solvent was removed by evaporation. The residue was chromatographed on silica gel using CHCl<sub>3</sub> or CHCl<sub>3</sub>–acetone 100:1 as eluent, and crystallized from MeOH.

*3,4,4',5'-Tetrahydro-5'-(hydroxyimino)-4'-(4-methylphenyl)-2'-(phenylimino)spiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5a, C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (979 mg, 46%); m.p.: 242–243 °C; <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.53 (s, 1H, NOH), 7.67 (d, 1H, <sup>3</sup>*J* = 7.8 Hz, CH arom), 7.44–7.37 (m, 3H, CH arom), 7.29 (t, 1H, <sup>3</sup>*J* = 7.3 Hz, CH arom), 7.19–7.12 (m, 2H, CH arom), 7.00 (d, 2H, <sup>3</sup>*J* = 8.0 Hz, CH arom), 6.82 (d, 2H, <sup>3</sup>*J* = 7.3 Hz, CH arom), 6.73 (d, 2H, <sup>3</sup>*J* = 8.0 Hz, CH arom), 4.84 (s, 1H, HC-4'), 2.71 (dt, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J*  $\approx$  5.5 Hz, CH<sub>2</sub>), 2.54 (ddd, 1H, <sup>2</sup>*J* = 15.0, <sup>3</sup>*J* = 11.0, 5.8 Hz, CH<sub>2</sub>), 2.23 (s, 3H, CH<sub>3</sub>), 2.02 (dt, 1H, <sup>2</sup>*J* = 15.0, <sup>3</sup>*J*  $\approx$  5 Hz, CH<sub>2</sub>), 1.88 (ddd, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J* = 11.0, 4.6 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 206.6 (C=O), 170.1, 152.1, 150.5, 137.7, 137.4, 137.0, 130.0, 129.9, 129.6, 129.3, 128.2, 127.4, 127.2, 127.0, 124.9, 118.8, 71.0, 60.9, 38.8, 25.9, 20.5 ppm; IR (KBr):  $\bar{\nu}$  = 3,183 (OH), 3,064, 3,023, 2,960–2,856 (C–H), 1,712 (C=O), 1,610, 1,591, 1,512, 1,488, 1,447, 1,380, 1,210, 1,121, 959 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 426 (100) [M]<sup>+</sup>, 276.7 (56), 260.7 (42), 259.8 (37), 231.9 (34), 115.2 (36), 91 (32).

*4'-(4-Chlorophenyl)-3,4,4',5'-tetrahydro-5'-(hydroxyimino)-2'-(phenylimino)spiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5b, C<sub>25</sub>H<sub>19</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (1.071 g, 48%); m.p.: 241–243 °C; <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.62 (s, 1H, NOH), 7.69 (d, 1H, <sup>3</sup>*J* = 8.0 Hz, CH arom), 7.45–7.37 (m, 3H, CH arom), 7.33–7.27 (m, 3H, CH arom), 7.17 (dd, 2H, <sup>3</sup>*J*  $\approx$  7, 7 Hz, CH arom), 6.87–6.81 (m, 4H, CH arom), 4.95 (s, 1H, HC-4'), 2.77 (dt, 1H, <sup>2</sup>*J* = 15.8, <sup>3</sup>*J*  $\approx$  5.5 Hz, CH<sub>2</sub>), 2.62 (ddd, 1H, <sup>2</sup>*J* = 14.9, <sup>3</sup>*J* = 11.3, 5 Hz, CH<sub>2</sub>), 2.11 (dt, 1H, <sup>2</sup>*J* = 14.9, <sup>3</sup>*J*  $\approx$  5 Hz, CH<sub>2</sub>), 1.92 (ddd, 1H, <sup>2</sup>*J* = 15.8, <sup>3</sup>*J* = 11.3, 4.4 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 206.3, 169.4, 151.7, 150.4, 137.7, 137.0, 132.5, 132.1, 132.0, 129.7, 129.3, 127.7, 127.5, 127.3, 127.1, 125.0, 118.8, 71.0, 59.9, 38.7, 26.0 ppm; IR (KBr):  $\bar{\nu}$  = 3,201 (OH), 3,026, 2,959–2,854 (C–H), 1,711 (C=O), 1,609, 1,592, 1,490, 1,383, 1,127, 1,088, 960 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 448 (38) [M + 2]<sup>+</sup>, 446 (100) [M]<sup>+</sup>, 313 (17), 311 (56), 282 (13), 266 (15), 247 (42), 246 (28), 218 (12), 116.1 (12), 115.1 (17).

*3,4,4',5'-Tetrahydro-5'-(hydroxyimino)-2'-[(4-methylphenyl)imino]-4'-phenylspiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5c, C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>S)*

Colorless crystals (1.003 g, 47%); m.p.: 238–240 °C; <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.55 (s, 1H, NOH), 7.66 (d, 1H, <sup>3</sup>*J* = 7.9 Hz, CH arom), 7.42 (t, 1H, <sup>3</sup>*J* = 7.4 Hz, CH arom), 7.31–7.16 (m, 6H, CH arom), 7.12 (d, 1H, <sup>3</sup>*J* = 7.4 Hz, CH arom), 6.83 (d, 2H, <sup>3</sup>*J* = 7.3 Hz, CH arom), 6.73 (d, 2H, <sup>3</sup>*J* = 8.2 Hz, CH arom), 4.86 (s, 1H, HC-4'), 2.69 (dt, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J*  $\approx$  5.5 Hz, CH<sub>2</sub>), 2.54 (ddd, 1H, <sup>2</sup>*J* = 15.0, <sup>3</sup>*J* = 11.2, 5.8 Hz, CH<sub>2</sub>), 2.29 (s, 3H, CH<sub>3</sub>), 2.01 (dt, 1H, <sup>2</sup>*J* = 15.0, <sup>3</sup>*J*  $\approx$  5 Hz, CH<sub>2</sub>), 1.82 (ddd, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J* = 11.2, 4.6 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 206.6 (C=O), 169.5, 152.1, 148.0, 137.7, 137.4, 134.2, 133.0, 130.2, 129.7, 129.6, 127.7, 127.6, 127.4, 127.2, 127.0, 118.8, 71.0, 61.1, 38.8, 25.9, 20.4 ppm; IR (KBr):  $\bar{\nu}$  = 3,205 (OH), 3,064, 3,027, 2,960–2,843 (C–H), 1,716 (C=O), 1,598, 1,504, 1,449, 1,391, 1,229, 1,209, 1,124, 1,069, 957 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 426 (100) [M]<sup>+</sup>, 277 (62), 261 (44), 260 (39), 248 (15), 232 (36), 218 (13), 217 (13), 116 (30), 115 (35), 91 (32).

*4'-(4-Chlorophenyl)-3,4,4',5'-tetrahydro-5'-(hydroxyimino)-2'-[(4-methylphenyl)imino]spiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5d, C<sub>26</sub>H<sub>21</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Crystallized from CHCl<sub>3</sub>. Colorless crystals of **5d**·CHCl<sub>3</sub> (949 mg, 33%); m.p.: 222–224 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.59 (s, 1H, NOH), 8.31 (s, CHCl<sub>3</sub>) 7.66 (d, 1H, <sup>3</sup>*J* = 7.8 Hz, CH arom), 7.42 (dt, 1H, <sup>3</sup>*J* = 7.3, <sup>4</sup>*J* = 1.3 Hz, CH arom), 7.33–7.25 (m, 3H, CH arom), 7.22–7.13 (m, 3H, CH arom), 6.85 (d, 2H, <sup>3</sup>*J* = 8.6 Hz, CH arom), 6.73 (d, 2H, <sup>3</sup>*J* = 8.2 Hz, CH arom), 4.93 (s, 1H, HC-4'), 2.76 (ddd, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J*  $\approx$  5.7, 5.0 Hz, CH<sub>2</sub>), 2.61 (ddd, 1H, <sup>2</sup>*J* = 14.9, <sup>3</sup>*J* = 11.4, 5.7 Hz, CH<sub>2</sub>), 2.29 (s, 3H, CH<sub>3</sub>), 2.10 (dt, 1H, <sup>2</sup>*J* = 14.9, <sup>3</sup>*J*  $\approx$  5 Hz, CH<sub>2</sub>), 1.91 (ddd, 1H, <sup>2</sup>*J* = 15.7, <sup>3</sup>*J* = 11.4, 4.6 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 206.3, 168.9, 151.8, 147.9, 137.7, 137.1, 134.2, 132.5, 132.1, 132.0, 129.7, 127.7, 127.5, 127.3, 127.1, 118.8, 79.1 (CHCl<sub>3</sub>), 71.0, 59.9, 38.7, 26.0, 20.4 ppm; IR (KBr):  $\bar{\nu}$  = 3,192, 3,062, 3,021, 2,917, 2,854, 1,713, 1,605, 1,493, 1,414, 1,398, 1,228, 1,216, 1,129, 1,091, 1,017, 997, 960 cm<sup>-1</sup>; ESI-MS: *m/z* (%) = 462.9 (39), 461.2 (100) [M + H]<sup>+</sup>, 376.5 (20).

*2'-[(4-Chlorophenyl)imino]-3,4,4',5'-tetrahydro-5'-(hydroxyimino)-4'-phenylspiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5e, C<sub>25</sub>H<sub>19</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Pale yellow crystals (1.092 g, 49%); m.p.: 240–242 °C; <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.62 (s, 1H, NOH), 7.69 (d, 1H, <sup>3</sup>*J* = 8.0 Hz, CH arom), 7.47–7.39 (m, 3H, CH arom), 7.30 (ddd, 1H, <sup>3</sup>*J* = 7.5, 7.5, <sup>4</sup>*J* = 1.1 Hz, CH



arom), 7.25 (t, 1H,  $^3J = 7.4$  Hz, CH arom), 7.19 (dd, 2H,  $^3J \approx 7.4, 7.4$  Hz, CH arom), 7.12 (d, 1H,  $^3J = 7.5$  Hz, CH arom), 6.88–6.81 (m, 4H, CH arom), 4.90 (s, 1H, HC-4'), 2.70 (dt, 1H,  $^2J = 15.6, ^3J \approx 5.5$  Hz, CH<sub>2</sub>), 2.55 (ddd, 1H,  $^2J = 15.2, ^3J = 11.1, 5.5$  Hz, CH<sub>2</sub>), 2.00 (dt, 1H,  $^2J = 15.2, ^3J \approx 5$  Hz, CH<sub>2</sub>), 1.83 (ddd, 1H,  $^2J = 15.6, ^3J = 11.1, 4.6$  Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 206.4$  (C=O), 171.2, 151.6, 149.3, 137.7, 137.0, 132.9, 130.1, 129.6, 129.3, 129.1, 127.7, 127.6, 127.4, 127.3, 127.0, 120.7, 71.1, 61.1, 38.8, 25.9 ppm; IR (KBr):  $\bar{\nu} = 3,233$  (OH), 3,065, 3,031, 2,963–2,844 (C–H), 1,718 (C=O), 1,607, 1,484, 1,451, 1,398, 1,230, 1,211, 1,125, 1,008, 959 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 448 (38) [M + 2]<sup>+</sup>, 446 (100) [M]<sup>+</sup>, 282 (13), 281 (33), 280 (15), 277 (66), 260 (13), 248 (15), 246 (12), 232 (28), 218 (17), 217 (17), 116 (23), 115 (29).

*2'-[(4-Chlorophenyl)imino]-3,4,4',5'-tetrahydro-5'-(hydroxyimino)-4'-(4-methylphenyl)spiro[naphthalene-1(2H),3'(2'H)-thiophene]-2-one (5f, C<sub>26</sub>H<sub>21</sub>ClN<sub>2</sub>O<sub>2</sub>S)*

Colorless crystals (1.377 g, 60%); m.p.: 216–218 °C; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 11.60$  (s, 1H, NOH), 7.67 (d, 1H,  $^3J = 7.9$  Hz, CH arom), 7.48–7.38 (m, 3H, CH arom), 7.29 (dd, 1H,  $^3J \approx 7.4, 7.4$  Hz, CH arom), 7.13 (d, 1H,  $^3J = 7.4$  Hz, CH arom), 6.99 (d, 2H,  $^3J = 8.0$  Hz, CH arom), 6.86 (d, 2H,  $^3J = 8.6$  Hz, CH arom), 6.72 (d, 2H,  $^3J = 8.0$  Hz, CH arom), 4.85 (s, 1H, HC-4'), 2.69 (ddd, 1H,  $^2J = 15.5, ^3J \approx 5.7, 5.3$  Hz, CH<sub>2</sub>), 2.55 (ddd, 1H,  $^2J = 14.9, ^3J = 10.8, 5.7$  Hz, CH<sub>2</sub>), 2.23 (s, 3H, CH<sub>3</sub>), 2.01 (dt, 1H,  $^2J = 14.9, ^3J \approx 5$  Hz, CH<sub>2</sub>), 1.88 (ddd, 1H,  $^2J = 15.5, ^3J = 10.8, 4.7$  Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 207.1, 171.9, 152.4, 149.9, 138.3, 137.8, 137.6, 130.6, 130.4, 130.2, 129.9, 129.7, 128.9, 128.1, 127.9, 127.7, 121.4, 71.7, 61.5, 26.6, 21.1$  ppm; IR (KBr):  $\bar{\nu} = 3,200, 3,054, 3,027, 2,912, 2,848, 1,719, 1,608, 1,513, 1,483, 1,444, 1,229, 1,161, 1,136, 1,091, 999, 965$  cm<sup>-1</sup>; ESI-MS: *m/z* (%) = 483.1 (11) [M + Na]<sup>+</sup>, 463.1 (39), 461.4 (100) [M + H]<sup>+</sup>.

*General procedure for synthesis of 6a–6f and 7a–7f*

To a solution of the appropriate iminothiophene **4a–4f** or **5a–5f** (0.5 mmol) in MeOH (for **4a–4f**: 80 cm<sup>3</sup>; for **5a–5f**: 120–160 cm<sup>3</sup>) at reflux temperature 1 cm<sup>3</sup> conc. HCl was added in one portion. For **5a–5f**, the mixture was gently heated under reflux for 0.5 h, then 1 cm<sup>3</sup> conc. HCl was added again. All the mixtures were stirred at reflux temperature for 3 h then concentrated to a volume of ca. 40 cm<sup>3</sup>. Then 40 cm<sup>3</sup> water was added, the mixture was left overnight, then most of the solution was decanted. The oily residue was dissolved in CHCl<sub>3</sub>, washed with plenty of water, and the organic layer was dried with MgSO<sub>4</sub>, concentrated, and purified by column

chromatography on silica gel using CHCl<sub>3</sub> as eluent. The products **6d**, **6f**, **7a**, **7d**, and **7f** crystallized from MeOH solution in a fridge.

*2-[2,5-Dihydro-4-(4-methylphenyl)-5-oxo-1-phenyl-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (6a, C<sub>26</sub>H<sub>21</sub>NO<sub>3</sub>S)*

Brown thick oil (55 mg, 26%); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.56$ –7.33 (m, 10H, CH arom), 7.28–7.24 (m, 1H, CH arom), 7.09 (d, 2H,  $^3J = 8.1$  Hz, CH arom), 3.54–3.37 (2 × d, AB system, 2H,  $^2J = 16.1$  Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>), 2.30 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 199.8, 172.2, 171.0, 140.6, 140.0, 133.9, 133.0, 131.6, 131.55, 131.0, 130.9, 130.0, 129.55, 129.4, 129.1, 128.7, 128.4, 127.3, 125.8, 51.9, 39.6, 21.5$  ppm; IR (KBr):  $\bar{\nu} = 3,439, 3,030, 2,948, 2,923, 2,854, 1,732, 1,710, 1,597, 1,499, 1,434, 1,380, 1,294, 1,222, 1,148, 1,016$  cm<sup>-1</sup>; ESI-MS: *m/z* (%) = 450.1 (42) [M + Na]<sup>+</sup>, 428.5 (70) [M + H]<sup>+</sup>, 412.5 (71), 396.6 (27).

*2-[4-(4-Chlorophenyl)-2,5-dihydro-5-oxo-1-phenyl-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (6b, C<sub>25</sub>H<sub>18</sub>ClNO<sub>3</sub>S)*

Dark green, thick oil (51 mg, 23%); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.57$ –7.43 (m, 6H, CH arom), 7.42–7.35 (m, 4H, CH arom), 7.30–7.21 (m, 3H, CH arom), 3.51–3.37 (2 × d, AB system, 2H,  $^2J = 15.6$  Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 199.3, 171.9, 170.9, 140.9, 136.5, 133.8, 133.0, 131.3, 131.2, 131.0, 130.7, 130.5, 129.9, 129.2, 129.0, 128.9, 128.4, 127.5, 127.1, 52.0$  (OCH<sub>3</sub>), 39.6 (CH<sub>2</sub>) ppm; IR (KBr):  $\bar{\nu} = 3,441, 3,063, 2,948, 2,852, 1,733, 1,592, 1,496, 1,379, 1,293, 1,223, 1,150, 1,092, 1,013$  cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 449.1 (40) [M + 2]<sup>+</sup>, 447.1 (100) [M]<sup>+</sup>, 418.1 (19), 416.1 (14), 414.1 (21), 390.1 (24), 389.1 (23), 388.1 (65), 376.1 (22), 374.0 (56), 346.1 (15), 221.0 (17), 189.2 (25).

*2-[2,5-Dihydro-1-(4-methylphenyl)-5-oxo-4-phenyl-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (6c, C<sub>26</sub>H<sub>21</sub>NO<sub>3</sub>S)*

Brown thick oil (23 mg, 11%); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.56$  (dd, 2H,  $^3J = 8.3, ^4J = 1.7$  Hz, CH arom), 7.48–7.25 (m, 11H, CH arom), 3.54–3.36 (2 × d, AB system, 2H,  $^2J \approx 16$  Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>), 2.42 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 199.8, 172.1, 170.9, 140.7, 138.9, 133.0, 131.53, 131.37, 131.25, 131.0, 130.9, 130.0, 129.9, 129.6, 128.7, 128.6, 128.1, 127.3, 51.9, 39.6, 21.3$  ppm; IR (KBr):  $\bar{\nu} = 3,453, 3,059, 2,948, 2,922, 2,853, 1,731, 1,710, 1,514, 1,444, 1,434, 1,382, 1,293, 1,148, 1,019$  cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 427.1 (100) [M]<sup>+</sup>, 398.1 (19), 394.1 (19), 369.1 (18), 368.1 (62), 367.1 (25), 355.1 (16), 354.1 (57), 326.1 (15), 221.0 (14), 191.1 (12), 189.1 (21).

2-[4-(4-Chlorophenyl)-2,5-dihydro-1-(4-methylphenyl)-5-oxo-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (**6d**, C<sub>26</sub>H<sub>20</sub>ClNO<sub>3</sub>S)

Dark green crystals (46 mg, 20%); m.p.: 135 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 7.53 (d, 2H, <sup>3</sup>J = 8.9 Hz, CH arom), 7.49–7.20 (m, 10H, CH arom), 3.50–3.37 (2 × d, AB system, 2H, <sup>2</sup>J = 15.5 Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>), 2.42 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ = 199.4, 172.0, 170.9, 140.8, 139.0, 136.4, 133.0, 131.32, 131.17, 131.10, 130.7, 130.5, 129.9, 129.8, 128.9, 128.1, 127.4, 127.2, 51.9, 39.6, 21.3 ppm; IR (KBr):  $\bar{\nu}$  = 3,433, 3,067, 2,950, 2,921, 2,853, 1,725, 1,587, 1,513, 1,481, 1,382, 1,308, 1,291, 1,263, 1,223, 1,148, 1,090, 1,013 cm<sup>-1</sup>; ESI-MS: *m/z* (%) = 484.3 (81) [M + Na]<sup>+</sup>, 464.3 (40), 462.4 (100) [M + H]<sup>+</sup>.

2-[1-(4-Chlorophenyl)-2,5-dihydro-5-oxo-4-phenyl-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (**6e**, C<sub>25</sub>H<sub>18</sub>ClNO<sub>3</sub>S)

Dark green solid (61 mg, 27%); m.p.: 77–79 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 7.57–7.23 (m, 13H, CH arom), 3.47–3.35 (2 × d, AB system, 2H, <sup>2</sup>J = 15.5 Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ = 199.4, 171.9, 170.9, 141.0, 134.8, 133.0, 132.3, 131.7, 131.2, 130.9, 130.3, 130.1, 129.81, 129.77, 129.5, 128.7, 128.5, 127.4, 52.0, 39.6 ppm; IR (KBr):  $\bar{\nu}$  = 3,448, 3,063–2,996, 2,948, 2,945, 1,736, 1,493, 1,434, 1,383, 1,305, 1,289, 1,270, 1,211, 1,159, 1,089, 1,018 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 449.1 (41) [M + 2]<sup>+</sup>, 447.1 (100) [M]<sup>+</sup>, 418.1 (23), 416.1 (14), 414.1 (22), 390.1 (31), 389.0 (31), 388.1 (86), 387.1 (29), 376.1 (24), 374.0 (63), 355.1 (12), 311.1 (13), 310.1 (18), 221.1 (20), 191.1 (15), 189.1 (28), 165.1 (13).

2-[1-(4-Chlorophenyl)-2,5-dihydro-4-(4-methylphenyl)-5-oxo-2-thioxo-1H-pyrrol-3-yl]benzeneacetic acid methyl ester (**6f**, C<sub>26</sub>H<sub>20</sub>ClNO<sub>3</sub>S)

Brown crystals (68 mg, 29%); m.p.: 109–110 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 7.52–7.33 (m, 9H, CH arom), 7.27–7.22 (m, 1H, CH arom), 7.10 (d, 2H, *J* = 8.1 Hz, CH arom), 3.48–3.35 (2 × d, AB system, 2H, <sup>2</sup>J = 15.5 Hz, CH<sub>2</sub>, + s, 3H, OCH<sub>3</sub>), 2.31 (s, 3H, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ = 199.5, 171.9, 170.9, 140.8, 140.2, 134.6, 133.0, 132.3, 131.6, 131.4, 131.1, 130.9, 130.0, 129.74, 129.65, 129.45, 129.37, 127.3, 125.7, 51.9, 39.6, 21.5 ppm; IR (KBr):  $\bar{\nu}$  = 3,445, 3,095–2,994, 2,948, 2,923, 1,734, 1,612, 1,493, 1,386, 1,305, 1,289, 1,270, 1,211, 1,160, 1,089, 1,016 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 463.1 (43) [M + 2]<sup>+</sup>, 461.1 (100) [M]<sup>+</sup>, 432.1 (17), 430.0 (16), 428.1 (28), 404.1 (23), 403.0 (25), 402.1 (59), 401.0 (27), 390.0 (18), 388.0 (44), 291.1 (21).

2-[2,5-Dihydro-4-(4-methylphenyl)-5-oxo-1-phenyl-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (**7a**, C<sub>27</sub>H<sub>23</sub>NO<sub>3</sub>S)

Brown crystals (91 mg, 41%); m.p.: 136–137 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ = 7.52 (dd, 2H, <sup>3</sup>J = 7.6, 7.6 Hz, CH arom), 7.48–7.39 (m, 6H, CH arom), 7.34–7.28 (m, 2H, CH arom), 7.22 (d, 1H, <sup>3</sup>J = 7.5 Hz, CH arom), 7.09 (d, 2H, <sup>3</sup>J = 8.5 Hz, CH arom), 3.60 (s, 3H, OCH<sub>3</sub>), 2.80–2.66 (2 × ddd, 2H, <sup>2</sup>J = 14.6, <sup>3</sup>J = 9.7, 9.5, 6.3 Hz, CH<sub>2</sub>), 2.42 (ddd, 1H, <sup>2</sup>J = 16.0, <sup>3</sup>J = 9.5, 6.3 Hz, CH<sub>2</sub>), 2.36–2.27 (m, 4H, <sup>3</sup>J = 9.7, 6.5 Hz, H-CH + s, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ = 200.3 (C=S), 173.0, 172.2, 140.6, 140.5, 139.0, 134.0, 131.3, 131.1, 130.8, 129.8, 129.5, 129.4, 129.1, 129.0, 128.7, 128.4, 126.4, 126.0, 51.6 (OCH<sub>3</sub>), 34.4 (C-3), 28.5 (C-2), 21.5 (CH<sub>3</sub>) ppm; IR (nujol):  $\bar{\nu}$  = 1,727 (C=O), 1,591, 1,278, 1,143 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 441.0 (100) [M]<sup>+</sup>, 412.0 (13), 408.1 (52), 368 (32), 367.0 (30), 354.0 (51), 348.1 (41), 326.0 (18), 320.1 (18), 203.0 (15), 202.0 (20).

2-[4-(4-Chlorophenyl)-2,5-dihydro-5-oxo-1-phenyl-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (**7b**, C<sub>26</sub>H<sub>20</sub>ClNO<sub>3</sub>S)

Brown thick oil (87 mg, 38%); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ = 7.56–7.40 (m, 8H, CH arom), 7.35–7.29 (m, 2H, CH arom), 7.26 (d, 2H, <sup>3</sup>J = 8.7 Hz, CH arom), 7.21 (d, 1H, <sup>3</sup>J = 7.4 Hz, CH arom), 3.60 (s, 3H, OCH<sub>3</sub>), 2.79–2.64 (2 × ddd, 2H, <sup>2</sup>J = 14.7, <sup>3</sup>J = 9.5, 9.1, 6.6 Hz, CH<sub>2</sub>), 2.46 (ddd, 1H, <sup>2</sup>J = 16.3, <sup>3</sup>J = 9.1, 6.4 Hz, CH<sub>2</sub>), 2.36 (ddd, 1H, <sup>2</sup>J = 16.3, <sup>3</sup>J = 9.5, 6.6 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ = 199.8 (C=S), 172.9, 171.9, 141.5, 138.9, 136.4, 133.8, 131.1, 130.6, 130.1, 129.7, 129.1, 129.05, 128.9, 128.8, 128.4, 127.3, 126.5, 51.6 (OCH<sub>3</sub>), 34.4 (C-3), 28.4 (C-2) ppm; IR (nujol):  $\bar{\nu}$  = 1,733 (C=O), 1,558, 1,492, 1,296, 1,166, 1,150, 1,090 cm<sup>-1</sup>; MS (70 eV): *m/z* (%) = 463.1 (40) [M + 2]<sup>+</sup>, 461 (100) [M]<sup>+</sup>, 430.1 (21), 428.1 (35), 388.0 (31), 376.0 (23), 374.0 (59), 372.1 (21), 368.1 (30), 340.1 (15), 202.1 (29), 93.0 (16), 77.0 (17).

2-[2,5-Dihydro-1-(4-methylphenyl)-5-oxo-4-phenyl-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (**7c**, C<sub>27</sub>H<sub>23</sub>NO<sub>3</sub>S)

Brown thick oil (70 mg, 32%); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ = 7.53 (d, 2H, <sup>3</sup>J = 8.6 Hz, CH arom), 7.42 (dt, 1H, <sup>3</sup>J = 7.6, <sup>4</sup>J = 1.4 Hz, CH arom), 7.35–7.25 (m, 9H, CH arom), 7.22 (d, 1H, <sup>3</sup>J = 7.5 Hz, CH arom), 3.59 (s, 3H, OCH<sub>3</sub>), 2.75 (ddd, 1H, <sup>2</sup>J = 14.6, <sup>3</sup>J = 9.8, 6.2 Hz, CH<sub>2</sub>), 2.67 (ddd, 1H, <sup>2</sup>J = 14.6, <sup>3</sup>J = 9.5, 6.4 Hz, CH<sub>2</sub>), 2.46–2.39 (m, 4H, <sup>2</sup>J = 16.1, <sup>3</sup>J = 9.5, 6.2 Hz, H-CH + s, CH<sub>3</sub>), 2.27 (ddd, 1H, <sup>2</sup>J = 16.1, <sup>3</sup>J = 9.8, 6.4 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):

$\delta = 200.3$  (C=S), 173.0, 172.1, 141.3, 138.9, 138.8, 131.2, 130.86, 130.82, 129.94, 129.86, 129.8, 129.5, 128.95, 128.88, 128.5, 128.1, 126.3, 51.6 (OCH<sub>3</sub>), 34.4 (C-3), 28.4 (C-2), 21.3 (CH<sub>3</sub>) ppm; IR (nujol):  $\bar{\nu} = 1,733$  (C=O), 1,558, 1,492, 1,296, 1,166, 1,150, 1,090 cm<sup>-1</sup>; MS (70 eV):  $m/z$  (%) = 441 (100) [M]<sup>+</sup>, 412.1 (14), 408.1 (42), 368.1 (33), 367.0 (23), 354.1 (55), 348.1 (39), 320.1 (16), 355.0 (32), 203.1 (17), 202.1 (16), 159.8 (34), 127.9 (49), 63.9 (49).

*2-[4-(4-Chlorophenyl)-2,5-dihydro-1-(4-methylphenyl)-5-oxo-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (7d, C<sub>27</sub>H<sub>22</sub>ClNO<sub>3</sub>S)*

Green crystals (52 mg, 22%); m.p.: 80–86 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.50$  (d, 2H, <sup>3</sup>J = 8.8 Hz, CH arom), 7.42 (ddd, 1H, <sup>3</sup>J ≈ 7.5, 7.5, <sup>4</sup>J = 1.5 Hz, CH arom), 7.36–7.17 (m, 9H, CH arom), 3.60 (s, 3H, OCH<sub>3</sub>), 2.81–2.62 (m, 2H, CH<sub>2</sub>), 2.52–2.29 (m, 5H, CH<sub>2</sub>, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 200.0$ , 173.0, 172.0, 141.4, 139.0, 138.9, 136.3, 131.13, 131.07, 130.65, 130.61, 130.1, 129.9, 129.8, 129.1, 128.9, 128.1, 127.4, 126.5, 51.7, 34.4, 28.4, 21.3 ppm; IR (KBr):  $\bar{\nu} = 3,434$ , 3,062, 2,921, 2,852, 1,726, 1,588, 1,514, 1,387, 1,293, 1,149, 1,092, 1,011 cm<sup>-1</sup>; ESI-MS:  $m/z$  (%) = 492.2 (100), 476.4 (25) [M + H]<sup>+</sup>, 460.4 (42), 444.5 (72), 442.5 (48).

*2-[1-(4-Chlorophenyl)-2,5-dihydro-5-oxo-4-phenyl-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (7e, C<sub>26</sub>H<sub>20</sub>ClNO<sub>3</sub>S)*

Brown thick oil (177 mg, 77%); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.55$ –7.46 (m, 4H, CH arom), 7.44–7.25 (m, 8H, CH arom), 7.21 (dd, 1H, <sup>3</sup>J = 7.6 Hz, <sup>4</sup>J = 1.4 Hz, CH arom), 3.59 (s, 3H, OCH<sub>3</sub>), 2.80–2.60 (2 × ddd, 2H, <sup>2</sup>J = 14.5, <sup>3</sup>J = 9.7, 9.2, 6.5 Hz, CH<sub>2</sub>), 2.42 (ddd, 1H, <sup>2</sup>J = 16.1, <sup>3</sup>J = 9.2, 6.5 Hz, CH<sub>2</sub>), 2.26 (ddd, 1H, <sup>2</sup>J = 16.1, <sup>3</sup>J = 9.7, 6.5 Hz, CH<sub>2</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 199.8$ , 173.0, 171.8, 141.6, 138.9, 134.7, 132.3, 131.4, 130.8, 130.6, 130.2, 129.9, 129.73, 129.68, 129.4, 129.0, 128.69, 128.63, 126.4, 51.7, 34.4, 28.4 ppm; IR (KBr):  $\bar{\nu} = 3,441$ , 3,059–2,994, 2,948, 2,850, 1,733, 1,493, 1,441, 1,377, 1,304, 1,285, 1,149, 1,089, 1,017 cm<sup>-1</sup>; ESI-MS:  $m/z$  (%) = 478.2 (100), 462.3 (40) [M + H]<sup>+</sup>, 446.3 (51), 430.4 (63), 428.4 (63).

*2-[1-(4-Chlorophenyl)-2,5-dihydro-4-(4-methylphenyl)-5-oxo-2-thioxo-1H-pyrrol-3-yl]benzenepropanoic acid methyl ester (7f, C<sub>27</sub>H<sub>22</sub>ClNO<sub>3</sub>S)*

Brown crystals (128 mg, 54%); m.p.: 88–91 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.51$ –7.25 (m, 9H, CH arom), 7.20 (dd, 1H, <sup>3</sup>J = 7.7 Hz, <sup>4</sup>J = 1.4 Hz, CH arom), 7.09 (d, 2H, <sup>3</sup>J = 8.0 Hz, CH arom), 3.58 (s, 3H, OCH<sub>3</sub>), 2.80–2.62 (m, 2H, CH<sub>2</sub>), 2.51–2.24 (m, 5H, CH<sub>2</sub>, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 200.0$ , 173.1, 172.0, 140.81,

140.78, 139.0, 134.6, 132.4, 131.4, 131.0, 130.8, 129.9, 129.8, 129.6, 129.5, 129.3, 129.1, 126.4, 125.9, 51.7, 34.4, 28.4, 21.5 ppm; IR (KBr):  $\bar{\nu} = 3,441$ , 3,090–2,985, 2,946, 2,917, 1,728, 1,608, 1,494, 1,385, 1,304, 1,290, 1,192, 1,150, 1,090, 1,017 cm<sup>-1</sup>; ESI-MS:  $m/z$  (%) = 514.2 (100), 498.2 (38) [M + Na]<sup>+</sup>, 476.4 (76) [M + H]<sup>+</sup>, 460.4 (63), 444.5 (63), 442.5 (83).

*General procedure for synthesis of 2-hydroxy-1H-indene-3-thiocarboxylic acid O-esters 16a, 16b*

A solution of **1a** (1 mmol) in 30 cm<sup>3</sup> dry MeOH (for **16a**) or EtOH (for **16b**) was heated under reflux until the starting material was consumed (TLC; 9 h for **16a**, 36 h for **16b**). The solvent was evaporated, and the residue was chromatographed on silica gel using CHCl<sub>3</sub> as eluent then crystallized from Et<sub>2</sub>O.

*2-Hydroxy-1H-indene-3-thiocarboxylic acid O-methyl ester (16a, C<sub>11</sub>H<sub>10</sub>O<sub>2</sub>S)*

Reddish brown crystals (89 mg, 43%); m.p.: 108–109 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 13.75$  (s, 1H, OH), 7.78 (d, 1H, <sup>3</sup>J = 7.8 Hz, CH arom), 7.32–7.23 (m, 2H, CH arom), 7.13 (ddd, <sup>3</sup>J ≈ 7.5, 7.5, <sup>4</sup>J = 1.2 Hz, CH arom), 4.26 (s, 3H, OCH<sub>3</sub>), 3.68 (s, 2H, H<sub>2</sub>C-3) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 205.4$ , 185.4, 140.3, 131.4, 127.1, 124.4, 123.5, 121.8, 114.6, 56.5, 39.3 ppm; IR (KBr):  $\bar{\nu} = 3,043$ , 3,001, 2,941, 2,919, 1,544, 1,483, 1,460, 1,440, 1,385, 1,304, 249, 1,234, 1,214, 1,183, 1,038, 944 cm<sup>-1</sup>; MS (70 eV):  $m/z$  (%) = 206.0 (37) [M]<sup>+</sup>, 174.0 (57), 146.0 (100), 145.0 (23), 134.0 (20), 102.0 (41).

*2-Hydroxy-1H-indene-3-thiocarboxylic acid O-ethyl ester (16b, C<sub>12</sub>H<sub>12</sub>O<sub>2</sub>S)*

Yellow crystals (135 mg, 61%); m.p.: 90–92 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 13.84$  (s, 1H, OH), 7.80 (d, 1H, <sup>3</sup>J = 7.8 Hz, CH arom), 7.33–7.23 (m, 2H, CH arom), 7.13 (ddd, <sup>3</sup>J ≈ 7.4, 7.4, <sup>4</sup>J = 1.1 Hz, CH arom), 4.74 (q, 2H, <sup>3</sup>J = 7.1 Hz, OCH<sub>2</sub>), 3.72 (s, 2H, H<sub>2</sub>C-3), 1.64 (t, 3H, <sup>3</sup>J = 7.1 Hz, CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 204.6$ , 185.4, 140.4, 131.4, 127.1, 124.3, 123.5, 121.7, 114.4, 66.1, 39.2, 13.9 ppm; IR (KBr):  $\bar{\nu} = 3,041$ , 2,988, 2,975, 2,892, 2,672, 2,629, 1,533, 1,483, 1,458, 1,366, 1,324, 1,300, 1,253, 1,211, 1,172, 1,150, 1,036, 1,022, 918 cm<sup>-1</sup>; MS (70 eV):  $m/z$  (%) = 220.1 (71) [M]<sup>+</sup>, 174.0 (100), 146.0 (80), 102.0 (25).

*X-ray structure analysis*

Compound **6d**, formula C<sub>26</sub>H<sub>20</sub>ClNO<sub>3</sub>S, crystallizes in the triclinic system, space group P-1, with unit cell parameters  $a = 8.3419(2)$ ,  $b = 11.3337(2)$ ,  $c = 13.8126(4)$  Å,  $\alpha = 66.627(1)$ ,  $\beta = 74.298(1)$ ,  $\gamma = 88.138(1)$  °,  $V = 1,149.86$

(3) Å<sup>3</sup>,  $Z = 2$ . A total of 4,478 independent reflections ( $R_{\text{int}} = 0.0204$ ) were collected from a sample (size  $0.35 \times 0.2 \times 0.15$  mm) using a Kappa CCD diffractometer and Mo-K $\alpha$  radiation. The structure was solved by direct methods with SHELXS97 [26] and refined by the full-matrix least-squares method on  $F^2$  using SHELXL97 [27] software. Final discrepancy indices for  $I > 2\sigma(I)$  were:  $R1 = 0.0476$ ,  $wR2 = 0.1048$ , and  $R1 = 0.0656$ ,  $wR2 = 0.1411$  for all data. The final difference Fourier map of electron density was featureless with the largest peak and hole of 0.18 and  $-0.23 \text{ eÅ}^{-3}$ , respectively. All calculations and molecular graphics were done using the WinGX package [28].

Compound **7a**, formula C<sub>27</sub>H<sub>23</sub>NO<sub>3</sub>S, crystallizes in the monoclinic system, space group P2<sub>1</sub>/c, with unit cell parameters  $a = 9.7300(1)$ ,  $b = 10.3974(2)$ ,  $c = 23.0241(4)$  Å,  $\beta = 95.731(1)^\circ$ ,  $V = 2317.63(6)$  Å<sup>3</sup>,  $Z = 4$ . A total of 4,046 independent reflections ( $R_{\text{int}} = 0.0246$ ) were collected from a sample (size  $0.35 \times 0.25 \times 0.15$  mm) using a Kappa CCD diffractometer and Mo-K $\alpha$  radiation. The structure was solved by direct methods using SHELXS97 [26] and refined by the full-matrix least-squares method on  $F^2$  using SHELXL97 [27] software. Final discrepancy indices for  $I > 2\sigma(I)$  were:  $R1 = 0.0526$ ,  $wR2 = 0.1413$ , and  $R1 = 0.0616$ ,  $wR2 = 0.1500$  for all data. The final difference Fourier map of electron density had the largest peak and hole of 0.63 and  $-0.408 \text{ eÅ}^{-3}$ , respectively. All calculations and molecular graphics were done using the WinGX package [28].

The structural data for compounds **6d** and **7a** have been deposited with Cambridge Crystallographic Data Centre as Supplementary Publications CCDC-720640 (**6d**) and CCDC-720639 (**7a**). Copies of these data can be obtained free of charge via [www.ccdc.cam.ac.uk/conts/retrieving.html](http://www.ccdc.cam.ac.uk/conts/retrieving.html) or from the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK; Fax: (internat.) +44-1223-336033; E-mail: [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk).

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